Sports Supplement Induces Dental Structure Corrosion: An in vitro Pilot Study

El Suplemento Deportivo Induce Corrosión en la Estructura Dental: Un Estudio Piloto in vitro

Lucas Thomazotti Berard1; Izabel Fernanda Machado2; João Pedro dos Santos Ferreira Moreira de Pinho3; Reinaldo Brito e Dias1 & Neide Pena Coto1


ABSTRACT: Dietary supplements are being consumed with an increasingly high frequency among sports practitioners, whether at professional and/or amateur level. The supplements contain some nutritional properties in their composition, so they can dissolve the hydroxyapatite crystals of the enamel and favor the process of dental corrosion. The objective was to measure the corrosive power of protein-based supplementation (Whey Protein), under conditions that resemble the use of the supplement by the athletes, increasing the ecological validity of the study. The teeth of the test group (TG) were placed in contact with the Whey protein solution and then exposed to artificial saliva. And the teeth of the control group (CG) were exposed only to artificial saliva. The analysis occurred in natural healthy molar teeth, so that each tooth of the TG was immersed in 50 mL of supplement for 1.5 minutes and then placed in contact with the artificial saliva for 30 seconds. The same procedure was performed 5 times a day for 30, 60, 90, 120, 150 and 180 days. Each group, in its time (TG0 to TG180), underwent analysis of superficial roughness with the aid of optical profilometer (TalySurf CC®, 3D model). The control group (CG) did not change its superficial roughness. Half of the teeth of the test group (TG) suffered loss of enamel surface. The values, in micrometers, of surface loss of the TG samples were 1.21; 2.1; 2.0; 1.04; 0.97; 0.8; 0.53; 1.14; 1.9; 2.0; 1.66; 1.80. The dietary supplement (Whey protein®) may be a potential cause of the dental corrosion process, considering the demineralization of hydroxyapatite that occurs along with the surface enamel loss.

KEY WORDS: dental etching, erosive potential, desmineralization.

INTRODUCTION

Dietary supplements have been ingested with an increasing frequency among the young and adult population, especially among sports practitioners (Bowman, 2002). This occurs due to the search for immediate results in their performance during physical activity (Middleman et al., 1998). The composition of dietary supplements impacts the athlete’s health, which may compromise their oral health, because some agents, such as Citrate, Glutamate, and Aspartate present in these substances are capable of removing Calcium from the tooth enamel, in addition to removing the Calcium present in the saliva. Thus, the remineralization capacity decreases, which enhances the process of dental corrosion (Amaechi et al., 1999; Wongkhantee et al., 2006; Barbour et al., 2008).

Flavoring agents present in the formula of dietary supplements, mainly citric acid, may be responsible for dental corrosion (Caims et al., 2002).

The term Corrosion refers to the chemical wear of the surface of the dental enamel. There are intrinsic (bulimia, gastro-oesophageal diseases) and extrinsic factors (supplement solutions, sports drinks, fruit juices, wines, soft drinks) that potentiate the corrosive process (West et al., 2003).

The process of dental corrosion presents multifactorial etiology, allowing chemical, biological and behavioral factors to influence the risk of corrosion (Nunn et al., 2003; Lussi, 2006; Lussi &

1 Department of Surgery, Prostheses and Maxillofacial Traumatology, School of Dentistry, University of São Paulo, São Paulo, Brazil.
2 Mechatronics and Mechanical System Engineering Department, Polytechnic School, University of São Paulo, São Paulo, Brazil.
3 Laboratory of biomechanics, School of Physical Education and Sports, University of São Paulo, São Paulo, Brazil.
A recent study by Vidal et al. (2017) evaluated the corrosive potential of Maltodextrin-based dietary supplement solutions, which is a glucose polymer commonly used in sports beverages. Maltodextrin is a carbohydrate that promises to keep blood glucose levels constant during sports practice and, consequently, to increase physical performance. This substance may be a potential causative agent of the corrosion process of dental enamel, but without submitting dental elements to cycling with acidic beverages.

Thus, the objective was to measure the corrosive power of protein-based supplementation (Whey Protein), under conditions that resemble the use of the supplement by the athletes, increasing the ecological validity of the study. The hypothesis is that the absence of maltodextrin will not cause a sharp decrease in the oral pH, not interfering in the formation of the acquired pellicle, nor causing dental corrosion.

MATERIAL AND METHOD

Experimental design. The dietary supplement Whey protein®Isofort was used for not having maltodextrin in its formula, because according to Vidal et al., supplements containing maltodextrin decrease salivary pH.

It is a longitudinal experiment with removal of experimental units without replacement.

pH. After dilution of the supplement studied, the solution was placed in a beaker and mixed with a magnetic stirrer (b10-b36). The pH of the solution was measured with a pH-meter (K39-1014B-Kasvi, São Paulo, SP, Brazil), (+ - 0.02). The value of 5.2 was obtained as a result of the pH measurement of the supplement studied.

Preparation of specimens. We selected 24 teeth that were exposed to the studied substance. Test groups (TG) were formed by groups of five teeth and evaluated every 30 days for 180 days total, and, for each analysis, 4 teeth were randomly removed.

Erosive challenge. Each TG tooth was immersed for 1.5 minutes in 50 mL of supplement. Then it was put in contact with the artificial saliva for 30 seconds. The same procedure was performed 5 times a day, with an interval of 24 hours between each cycle, and, during this interval, all samples were kept in artificial saliva. The saliva used followed the composition reported in Table I.

Table I. Saliva composition used in the study.

<table>
<thead>
<tr>
<th>Salts</th>
<th>Quantity (g/L)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Monopotassium phosphate</td>
<td>3.40</td>
</tr>
<tr>
<td>Sodium Phosphate Dibasic Anhydrous</td>
<td>3.41</td>
</tr>
<tr>
<td>Potassium Bicarbonate</td>
<td>15.02</td>
</tr>
<tr>
<td>Sodium Chloride</td>
<td>5.84</td>
</tr>
<tr>
<td>Magnesium Chloride</td>
<td>0.30</td>
</tr>
<tr>
<td>Calcium Chloride</td>
<td>2.21</td>
</tr>
<tr>
<td>Citric Acid</td>
<td>4.80</td>
</tr>
</tbody>
</table>

Optical profilometer surface analysis. Each group, in its time (TG to TG180), underwent analysis of surface roughness with the aid of optical profilometer (Talysurf CCI®, 3D model). Data were analyzed by ProScan Software®. The differences in height between the reference and the exposed area were quantified in micrometers (µm).

Each tooth had its vestibular surface measured 6 times, 3 times the region of 2 mm above (crown) the cervical line and 3 times the region of 1 mm below (root) the cervical line. Thus, the mean number of numerical findings was made to designate the values of the measured regions. The groups TG0, TG30, TG60, TG90, TG120, TG150, TG180 were formed.

Statistical procedures. A One-Sample T-test, with Bonferroni adjustment, was conducted to test the corrosion effects of the different time exposures. Therefore, the significance level was set at 0.008 (0.05/6). We compared the difference between initial and final period of the experimental unities with zero. All statistical tests were conducted on SigmaStat 3.5 (Systat Software, Inc).

RESULTS

The main results of our study can be summarized in the following panel (Fig. 1).

The superficial analysis, with the aid of the profilometer, allowed us to record roughness values using a 3D view scale with the height of the enamel.
prisms before the immersion of the specimens in the food supplement solution (Fig. 2). The profilometer also recorded the enamel surface of the same sample after the erosive challenge of 60 days, where one can observe the decrease in the height of the enamel prisms (Fig. 3).

Fig. 1. Mean and standard deviation of the dental surface erosion (in mm) differences between the initial period and the ith (T30 – T150) period.

Fig. 2. Surface of dental enamel in 3D view before immersion of a sample.

Fig. 3. Surface of dental enamel in 3D view after immersion of Sample 15 (EG60).
DISCUSSION

The objective of this research was to measure the corrosive power of dietary supplement without the presence of maltodextrin. This study presents ecological validity by submitting dental elements to erosive challenges similar to that performed by athletes during training and competitions.

In this investigation, the optical profilometer was used, which is a device that allows the topographic analysis of the surface of the most diverse materials. The superficial reading of the delineated area was performed with the laser beam emitted by the equipment, which also enables the translation into numerical data and values, in micrometers of the surface scan performed. It is an instrument of wide application, in several scales, which allowed the analysis of the degradation of the dental enamel surface and loss of the dental tissue, from the erosive challenges (Scaramucci et al., 2011), indicating thus the corrosive power of the supplement studied.

In the oral cavity, the teeth are protected by salivary film or acquired pellicle, composed of salivary glycoproteins and antibodies, through which the existence of remineralization effects is observed. The supplement, when ingested, may suffer buffering action, that is, the acidity of the beverage is minimized by the protein agents present in saliva (Amaecchi et al., 1999).

In TGs 150 and 180, a significant loss of surface roughness was observed, precisely due to the longer exposure time of the samples to the dietary supplement solution. Thus, the process of formation of the acquired pellicle was not able to curb the dental corrosion, so that the superficial loss generated in the dental enamel was more pronounced than the remineralization process. From the pH measurement, the value of 5.2 was verified. We also observed that the solution studied presents flavoring agents in its composition, such as Citric acid, Glutamic acid and Aspartic acid, which confer a more pleasant flavor to the supplement, but these acids contain chelating substances capable of generating the depletion of Calcium in saliva and dental enamel, contributing thus to the development of the corrosion process, as Zero in 1996 (Zero, 1996) and Cairns et al. in 2002 have shown.

Acidic beverages present different patterns of adhesion to the dental structure. We suggest that a longer time of exposure of the beverage to the tooth implies an increased risk of corrosion. This condition worsened by the fact that saliva cannot displace the film formed by the beverage on the dental surface, so that the acquired pellicle has its remineralization capacity decreased (Ireland et al., 1995; Jager et al., 2008).

The development of beverages with addition of Calcium and Phosphate ions was considered; however, we found that the addition of these ions confers a metallic flavor to the beverage (West et al., 2001). Researchers suggest that proteins adsorbed to enamel could modify the dissolution process of this substrate. Because of this property, the addition of proteins in acidic beverages could be a strategy to minimize corrosive wear (Ramalingam et al., 2005). Given this context, the incorporation of casein, protein from the group of phosphoproteins and found in milk, would imply the reduction in the dissolution rate of hydroxyapatite, regardless of the pH value of the acid substance (Jensdottir et al., 2005).

The injuries resulting from the ingestion of acidic beverages are more frequently found by vestibular in the cervical third of the anterior teeth, although there is the possibility of occurring in any region of the dental element. The cervical area is usually the most affected because the self-cleaning is lower than in other regions and, with this, the acid remains at this site for a longer period. Saliva does not act quickly in this location and its buffering effect takes longer to occur (Fuller & Johnson, 1977; Levitch et al., 1994). Also, the enamel layer of the cervical region is thinner, which favors the higher prevalence of non-curious lesions in this portion of the dental element (Boric et al., 2004). Thus, in this study, we investigated only the region that is more susceptible to loss of enamel structure, where there was no pattern of superficial enamel loss per region, since this loss could be observed in the three areas analyzed (mesiodistal, mesiovestibular, vestibular).

For an athlete, these deleterious effects are accentuated because there is a decrease in salivary flow and an increase in the concentration of total salivary protein, as a consequence of a state of dehydration during physical activity (Walsh et al., 2004). Thus, it is important that the dentist promotes the orientation of healthy practices among users of dietary supplements, whether professional athletes, amateurs or regular practitioners of physical activities, such as the decrease in the time in which they remain in contact.
with the dental elements and the reduction in the frequency of use of these beverages, especially among those individuals who have the habit of drinking other beverages considered acidic, such as soft drinks, fruit juices, energy and isotonic drinks (Abal et al., 2009). The primary consequences of the indiscriminate use of acidic beverages such as dietary supplement are dentin hypersensitivity and non-caries cervical lesion (Wongkhantee et al.). It is also noteworthy that brushing should not be performed immediately after ingestion of carbonated beverages, because these processes associated, by mechanical action, promote the fragilization of hydroxypatite crystals, increasing the corrosion (Devlin et al., 2006; Sundaram et al., 2007). In this sense, a previous mouthwash with an alkaline solution is recommended in an attempt to minimize the disorganization of enamel prisms (Eccles, 1979).

This pilot study sought to bring a realistic and ecological analysis of the nutritional habit of many athletes, and it should be considered for studies with greater robustness. However, it brings important clinical relevance, since it indicated a corrosive power of the supplement without maltodextrin.

The evidence from the results of this study suggests that the dietary supplement (Whey protein®) may be a potential cause of the dental corrosion process, considering the demineralization of hydroxypatite that occurs along with the enamel surface loss.


RESUMEN: Los suplementos dietéticos se consumen con una frecuencia cada vez más alta entre los practicantes de deportes, sea a nivel profesional y / o aficionado. Los suplementos contienen algunas propiedades nutricionales en su composición, por lo que pueden disolver los cristales de hidroxiapatita del esmalte y favorecer el proceso de corrosión dental. El objetivo fue medir el poder corrosivo de la suplementación a base de proteínas (proteína de suero), en condiciones que se asemejan al uso del suplemento por parte de los atletas, aumentando la validez ecológica del estudio. Los dientes del grupo de prueba (TG) se pusieron en contacto con la solución de proteína de suero y luego se expusieron a saliva artificial. Y los dientes del grupo de control (CG) estuvieron expuestos solo a saliva artificial. El análisis se realizó en dientes molares sanos naturales, cada diente del TG se sumergió en 50 ml de suplemento durante 1,5 minutos y luego se puso en contacto con la saliva artificial durante 30 segundos. El mismo procedimiento se realizó 5 veces al día durante 30, 60, 90, 120, 150 y 180 días. Cada grupo, en su momento (TG0 a TG180), se sometió a un análisis de rugosidad superficial con la ayuda de un perfilómetro óptico (Talysurf CCI®, modelo 3D). El grupo de control (CG) no cambió su rugosidad superficial. La mitad de los dientes del grupo de prueba (TG) sufrieron pérdida de la superficie del esmalte. Los valores, en micrómetros, de pérdida de superficie de las muestras de TG fueron 1,21; 2,1; 2,0; 1,04; 0,97; 0,8; 0,53; 1,14; 1,9; 2,0; 1,66; 1,80. El suplemento dietético (Whey protein®) puede ser una causa potencial del proceso de corrosión dental, considerando la desmineralización de la hidroxiapatita que ocurre junto con la pérdida de esmalte superficial.

PALABRAS CLAVE: grabado dental, potencial erosivo, desmineralización.

REFERENCES


Corresponding author:
Lucas Thomazotti Berard
Department of Maxillofacial Surgery
Prosthetics and Traumatology
School of Dentistry
University of São Paulo
Av. Prof. Lineu Prestes, 2227
Cidade Universitária, 05508-900
São Paulo – SP
BRAZIL

E-mail: lucas.berard@usp.br

Corresponding author:
Dr. Neide Pena Coto
Department of Surgery
Prostheses and Maxillofacial Traumatology
School of Dentistry
University of São Paulo
Avenida Lineu Prestes
2227 Cidade Universitária
São Paulo/SP
BRAZIL

Email: npcoto@usp.br

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